# The Significance of Your First Morning Urine pH and Its Proper Measurement

Your first morning urine pH gives a good indicator of the body's mineral reserve and its acid/alkaline state. The body routinely uses overnight rest time to excrete excess acids. This capacity varies based on toxin load and individual ability to make energy, to make toxins inactive, and to excrete them.

## How does one monitor the pH?

To test your pH, follow these simple steps:

- 1. Obtain a packet of pH (Hydrion<sup>TM</sup>) test paper with a test range of 5.5 to 8. Your local dispensary or pharmacist should have this item or be able to order it for you.
- 2. First thing in the morning, just before your first urination, open the test tape packet and cut off two or three inches of the paper tape. Now wet the test tape with urine. For best results, a 6-hour to 8-hour period of rest prior to pH testing is needed.
- 3. As the tape is moistened with urine it will change color. The color relates to the urine's acid or alkaline state and ranges from yellow to dark blue. Match the color of your test strip with the color chart on the back of the test tape packet.
- 4. Jot down the number that corresponds to the color of the urine-moistened tape has taken on. This can be done daily or periodically based on your needs as recommended by your physician.

### · How do I know if I am acidic or alkaline?

Any number below 7.0 means that your urine is on the acid side. The lower the number, the more acid the condition. For example, a number of 5.0 indicates 10 times more acidity than 6.0. A number of 7.0 indicates the neutral state, neither acid or alkaline. Ideally, your first morning urine pH should be in a pH range of 6.5 to 7.5. When your first morning urine is neutral or just slightly acidic, this indicates that your overall cellular pH is appropriately alkaline and that the small amounts of acids built up from normal metabolism have been easily concentrated for excretion. The cells of your body function best in an alkaline state.

### • What if my urine pH is below 6.5?

If your readings fall below 6.5, then you should begin changes aimed at alkalinizing your diet. Listed on page 13 are simple modifications that will help alkalinize your system. In the beginning, because of the acid-forming tendency of the standard American diet, you may well have low pH readings. Occasionally, you may find a 7.5 to 8.0 reading; this is acceptable. If your pH readings are consistently greater than 7.5, this occurrence represents a "false alkalinity" and indicates a catabolic state involving tissue breakdown.

# Guidelines for Developing Your Alkaline Diet

Examine the chart on page 14 entitled, Food and Chemical Effects of Acid/Alkaline Body Chemical Balance. If you are in an acid state, the first step is to eat more alkalinizing vegetables, fruits, spices, and lentils. Strive for two cups of alkalinizing vegetables at both lunch and dinner. Enjoying a breakfast of alkaline fruits or oatmeal while limiting high protein foods will also go a long way toward reducing your acidity. In addition, the following simple changes are especially helpful for quickly alkalinizing yourself:

- Drink the juice of one-half a lime, lemon, or a teaspoon of apple cider vinegar in 6-8 ounces of water
  a few times during the day. You may think this would make you more acidic. Not so! These substances
  metabolize to an alkaline residue due to the dicarboxylic acids that the body uses to make energy in a
  way that gives off an alkaline (bicarbonate) product. See the acid-alkaline effects on body chemistry chart
  on page 14 for more examples.
- Add lentils, ginger, yams, and sweet potatoes to your diet on a regular basis. These foods will give you
  an alkaline boost.
- Make it a point to eat daily at least two cups of alkalinizing greens (kale, mustard and turnip greens, collards, or endive). Lettuce is fine to eat, but not in place of alkaline greens. Grated daikon radish is a wonderful alkalinizing condiment.
- Add miso and seaweed to soups and other dishes as both a great digestive aid and as an alkalinizer.
- Eat more of the alkalinizing grains like oats, quinoa, and wild rice.
- Enjoy liberal amounts of fruits, especially watermelon.
- If you suffer from gas, bloating, or weak digestion, eat cooked fruit and small amounts of fresh juices.
- Certain supplements like a fully buffered ascorbate (vitamin C), soluble (ionized) magnesium, and
  L-Glutamine [with Pyridoxal-alpha-ketoglutarate (PAK)] will also alkalinize you and should be used
  in doses as needed based on your metabolism. We have found that an optimum dose of fully buffered
  ascorbate with a combination of calcium, magnesium, potassium, and zinc is a health-promoting way to
  alkalinize, energize, and enhance metabolism of toxins at the same time.

Be patient and persistent. Remember, your pH indicates your reserve of alkaline minerals. An alkaline pH indicates good reserves of the enzyme-activating alkaline minerals. It can take time to build up these reserves. It may have taken years to become depleted. Do not be discouraged with a slow movement toward the ideal alkaline state—a first morning pH of 6.5 to 7.5.

# The Importance of an Alkaline Diet

The internal environment of our bodies is maintained at a pH of just about 7.0. This means our internal environment is alkaline. Maintenance of this state is a dynamic, not static, process mediated moment to moment by numerous reactions that produce acid products. Our internal chemical equilibrium is primarily controlled by our lungs, kidneys, intestines, and skin. For necessary reactions and functions to occur, our body must maintain a proper pH. Adequate alkaline reserves are necessary for optimal pH adjustment. The body needs oxygen, water, and acid-buffering minerals to accomplish the pH buffering, while also briskly eliminating waste products.

When an alkaline environment is maintained in the body, metabolic, enzymatic, immunologic, and repair mechanisms function at their best. The acid-forming metabolics of stress and inflammation and of high fat and high protein foods are adequately and effectively neutralized only when sufficient mineral-buffering reserves are present. Mineral-buffering reserves are the gift that alkaline-forming foods give to our body. A diet that is predominantly alkaline forming is essential to the maintenance of sustained health.

Most vegetables and fruits contain higher proportions of alkaline-forming elements than other foods. These foods promote a more alkaline enviror, nent in the body. For example, commercial corn, batley, soybeans, and legumes are acid forming. This may reflect breeding selection in the last fifty years that favored higher carbohydrate and fat content. Traditional organically or biodynamically grown forms of these grains and grasses may well be much less acid forming. Surprisingly, despite their pronounced acid flavor, citrus fruit and rhubarb form alkaline residues. This is because their distinctive organic acids like citric, succinic, fumaric, and malic (Krebs' DCA or dicarboxylic acid) metabolize to water and alkalinizing bicarbonate, while producing energy (ATP) inside the cell.

Body balance, in terms of acid-alkaline state, is a pH of 7.450 for blood in the arteries and 7.350 for blood in the veins. Acid-alkaline equivalence is a pH of 7.000. Thus, a healthy body means a pH that is slightly alkaline. This means there are more buffering mineral receptors for electrons than acid-forming electron donors.

In foods containing large amounts of protein and fat, the acid-forming elements predominate over the alkaline-forming elements. Thus, cow's milk

and related dairy products are acid-forming, although goat and sheep milk/cheeses (with less fat and protein) produce less acid. The one dairy product exception is clarified butter (known as "ghee" in Indian cookery), which has alkalinizing short chain fats known as butyrates and caprylates. The butyrates and caprylates present in ghee are also thought to promote healthy bacterial growth in the intestines, promote repair of the intestine wall, and suppress pathogen growth of some yeasts and parasites if they are present.

Whole grains give an acid reaction disproportionate to their protein content due to the extra phosphorus present in the phytates. The phosphate content of commercial grains may be higher than traditional, organic, or biodynamic sources in part because of fertilizer differences and plant strain selection. Although most fruits have an alkaline effect, some such as prunes, plums, and cranberries make a net contribution of acid to the body since they contain organic acids that are not metabolized by the body. Nuts such as coconuts, almonds, and chestnuts are alkaline forming, while others like peanuts (a legume) and walnuts yield net acid. Highly refined and processed foods consisting chiefly of fats, sugars, and simple starches, along with protein-rich foods are metabolically acidifying.

The chart on the back of this page titled, Food & Chemical Effects on Acid/Alkaline Body Chemical Balance, presents the message that, in general, fruits, vegetables, lentils, seeds, sprouts, roots, and tubers are healthfully alkalinizing, while grains, grasses, fowl, fish, seafood, dairy products, meats, and most beans are acidifying. Here is a way to simplify this and make it memorable. If it comes from under or near the ground, it is likely to be alkalinizing. If it comes from on or high above the ground, it is likely to be acid forming.

The specifics of how each food was categorized on this chart are based on a formula wherein protein, fat, carbohydrate, mineral, and other specific factors were taken into account. More specifically, the basic neutral and acidic end-products of protein, fat, and carbohydrate digestion were calculated, and the content of minerals and special factors were also accounted. A computation was used to determine the relative degree of acid- or alkaline-forming effects of the food on body chemistry. Based on this determination, the items were placed in the appropriate acid or alkaline group on the chart.

Reference: Jaffe R and Donovan P. Your Health: A Professional User's Guide Sterling, Va.: Health Studies Collegium, 1993.

# Food & Chemical Effects on Acid / Alkaline Body Chemical Balance

Most Alkaline	More Alkaline	Low Alkaline	Lowest Alkaline	Food Category	Lowest Acid	Low Acid	More Acid	Most Acid
Baking Soda	Valerian Licorice  Black Cohash	Bergamot, Echinacea, Chrysanthemum, Ephedra, Fevefew, Goldenseal, Lemongrass	White whow bark Slippery Elm Artemesia Annua	spice/nero	Curry	vanilia Stevia	Numeg	rudingjanjeny
Sea Salt Mineral Water	• Kambucha		Sulfite Ginger Tea	Preservative	MSG Kong Coffee	Benzoate	Aspartame	Table Salt (NaCl)
			O	O		Black Tea		Yeast/Hops/Malt
	Molasses		• Sucanat	Sweetener	Honey/Maple Syrup	Polomia Vincent	Saccharin	Sugar/Cocoa
1 1 1 1 1 1	ouy sauce	Cluer Amegar	· Omeoosiii viilegai	Villegar	Rice Amegai	равание уневан		willie/Acenc villegal
• Umeboshi Plum		• Sake	<ul> <li>Algae, Blue-Green</li> </ul>	Iherapeutic		Antihistamines	Psychotropics	Antibiotics
			<ul> <li>Ghee (Clarified Butter)</li> </ul>	Processed Dairy	Cream/Butter	Cow Milk	<ul> <li>Casein, Milk Protein, Processed Cheese</li> <li>Cottage Cheese</li> </ul>	Processed Cheese
			Human Breast Milk	Cow/Human	Vogilit	Aged Cheese	New Cheese	Ice Cream
				Soy	d	Soy Cheese	Soy Milk	
				Goat/Sheep	Goat/Sheep Cheese	Goat Milk		
		Quail Egg	<ul> <li>Duck Egg</li> </ul>	Egg	Chicken Egg			
				Meat	Gelatin/Organs	Lamb/Mutton	Pork/Veal	Beef
				Game Fish/Shell Fish	• Venison Fish	Boar/Elk/•Game Meat Bear Shell Fish/Mollusks • Mu	Bear     Mussel/Squid	Lobster
				Fowl	Wild Duck	Goose/Turkey	Chicken	Pheasant
			Oat		Triticale	Buckwheat	Maize	Barley
			'Grain Coffee'	Grain	Millet	Wheat	Barley Groat	Processed Flour
			• Quinoa	Cereal	Kasha	• Spelt/Teff/Kamut	Corn	
			Japonica Rice	Grass	Brown Rice	White Rice	Oat Bran	
	Poppy Seed	Primrose Oil	Avocado Oil	Nut	Pumpkin Seed Oil	Almond Oil	Pistachio Seed	<ul> <li>Cottonseed Oil/Meal</li> </ul>
Pumpkin Seed	Cashew	Sesame Seed	Seeds (most)	Seed/Sprout	Grape Seed Oil	Sesame Oil	Chestnut Oil	Hazelnut
	Chestnut	Cod Liver Oil	Coconut Oil	Oii	Sunflower Oil	Safflower Oil	Lard	Walnut
	Pepper	Almond	Olive/Macadamia Oil		Pine Nut	Tapioca	Pecan	Brazil Nut
Hydrogenated Oil		• Sprout	Linseed/Flax Oil		Canola Oil	Seitan or Tofu	Palm Kernel Oil	Fried Food
Lentil	Kohlrabi	Potato/Bell Pepper	Brussel Sprout	,	Spinach	Split Pea	Green Pea	Soybean
Brocoflower	Parsnip/Taro	Mushroom/Fungi	Beet	Bean	Fava Bean	Pinto Bean	Peanut Poo	Carob
N W har have		Cahhaga	Calery/Scallion	AcSerable	Rlack-aved Pea	Navy/Red Rean	OHOW I Ca	
Onion/Miso	Kale/Parsley	Rutahaga	Okra/Cucumber	Legume	String/Wax Rean	Aduki Bean	Legumes (other)	
Daikon/     Taro Root	Endive/Arugula	<ul> <li>Salsify/</li></ul>	Turnip Greens	Pulse	Zucchini	Lima or Mung Bean	Carrot	
<ul> <li>Sea Vegetables (other)</li> </ul>	Mustard Greens		Squash	Root	Chutney	Chard	Chick Pea/Garbanzo	930 978
Burdock/• Lotus Root	Ginger Root		Lettuce		Rhubarb			
Sweet Potato/ rain	BIOCCOIL	Oreens	Jicama	0				
Nactarina		Dear	Apricot	Citrus Fruit	Coconut	Dhim	Cranharry	
Persimmon	Honeydew	ado	Banana		Pickled Fruit	Prune	Pomegranate	
			Blueberry		Dry Fruit	Tomato	23	
'n		Blackberry	Pineapple Juice	Fruit	Fig			
Tangerine	<ul> <li>Dewberry</li> </ul>	Cherry	Raisin, Currant		Persimmon Juice			
Pineapple	erry	Peach	Grape		• Cherimoya			
	Mango	Papaya	Strawberry		Date			

• Therapeutic, gournnet, or exotic items

Italicized items are NOT recommended.

Prepared by Dr. Russell Jaffe, Fellow, Health Studies Collegium. Reprints available from ELISA/ACT Biotechnologies, 14 Pidgeon Hill, #300, Sterling, VA 20165. Sources include USDA food data base (Rev 9 & 10), Food & Nutrition Encyclopedia; Nutrition Applied Personally, by M. Walczak; Acid & Alkaline by H.Aihara. Food growth, transport, storage, processing, preparation, combination, & assimilation influence effect intensity. Thanks to Hank Liers for his original work. [Rev 10/02]